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42 Chapter 2 Environmental Chemistry Section 2.2 Stoichiometry 43 We already know that there are 44 grams per mole of CO<sub>2</sub> so we do not need to recalculate that. Two moles of butane (2 mol x 58 g/mol = 116 g) yields 8 moles of carbon dioxide (8 mol x 44 g/mol = 352 g CO<sub>2</sub>) , So we can set up the following proportion: 116 g C<sub>4</sub>H<sub>10</sub> \_ 100 g C,H<sub>10</sub>

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