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STUDY GUIDE FOR CONTENT MASTERY Section 6.3 Periodic Trends In your textbook, read about atomic radius and ionic radius. Circle the letter of the choice that best completes the statement or answers the question. 1. Atomic radii cannot be measured directly because the electron cloud surrounding the nucleus does not have a clearly defined c. outer edge.

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STUDY GUIDE FOR CONTENT Section 4.4 Changes to the Nucleus—Nuclear Reactions In your textbook, read about radioactivity. For each item in Column A, write the letter of the matching item in Column B.

Column A

1. The rays and particles that are emitted by a radioactive material
2. A reaction that involves a change in an atom's nucleus
- 3.

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STUDY GUIDE FOR CONTENT MASTERY CHAPTER 17

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Reaction Rates Section 17.1 A Model for Reaction Rates

In your textbook, read about expressing reaction rates and explaining reactions and their rates. Use each of the terms below just once to complete the...

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This Study Guide for Content Mastery for Chemistry: Matter and Change will help you learn more easily from your textbook. Each textbook chapter has six study guide pages of questions and exercises for you to complete as you read the text. The study guide pages are divided into sections that match those in your text. You will find that the directions in the Study

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CHAPTER 12 STUDY GUIDE FOR CONTENT MASTERY
Section 12.1 continued In your textbook, read about mole ratios. Answer the questions about the following chemical reaction. sodium iron(III) oxide 0 sodium oxide iron 6Na(s) Fe 2 O 3 (s) 0 3Na 2 O(s) 2Fe(s ...

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test score ...

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STUDY GUIDE FOR CONTENT MASTERY Section 12.2 Stoichiometric Calculations In your textbook, read about mole-to-mole conversion. Read the following passage and then solve the problems. In the equation that follows each problem, write in the space provided the mole ratio that can be used to solve the problem.

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